cell cultures from mouse brain neurons and quantified dimerization of the BDNF receptor due to both direct and trans-activation via dopamine.

199-Pos

Why Is α4β2 Nachr More Sensitive to Volatile Anesthetics Than α7 Nachr?

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Two subtypes of neuronal nicotinic acetylcholine receptors (nAChRs) show different functional sensitivities to volatile anesthetics: the α4β2 nAChR is hypersensitive while the α7 nAChR is insensitive. To understand why these homologous proteins have different functional responses to volatile anesthetics, we performed multiple sets of 20-ns molecular dynamics (MD) simulations on the closed- and open-channel α7 in the absence and presence of halothane, and compared the results with those from a similar study on the α4β2 nAChR (Liu, et al. 2009). The details about construction of receptor structural models were published (Haddadian, et al. 2008). Initial halothane docking and subsequent MD simulations revealed several halothane-binding sites in α7. Consistent with observations from α4β2, free energy perturbation calculations showed that halothane had higher binding affinity in the closed- than the open-channel; α7 and α4β2 had a comparable number of high affinity sites. GNM analysis showed that halothane induced profound changes in correlated domain motion of the open-channel α4β2, especially between the Cys-loop and the TM2-TM3 linker, but had a negligible impact to the motion of the α7. Salt bridges between these loops in the β2 subunit may be responsible for the aforementioned observation. Flexibility of several key loops in the open-channel α4β2 changed considerably in the presence of halothane, but the same loops in the open-channel α7 showed little change. Taken together, our results suggest that halothane binding in nAChRs may be necessary, but not sufficient to produce essential dynamical changes that alter protein functions. Although α7 and α4β2 are homologous, specific residues in key loops may make α4β2 more susceptible to volatile anesthetics while α7 unaffected. Supported by NIH (R01GM66358, R01GM56257, and T32GM075770) and NCSA through the PSC.

200-Pos

Conformational Docking of Multiple Toxins Against Kv1-Channels Highlight Key Motifs For Selectivity

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Scorpions and other venomous predators are known to disable their prey via injection of peptides, which interfere with ion channels involved in neural signalling. These peptides are both highly specific blockers and potential scaffolds for toxin-based therapeutics of diseases such as multiple sclerosis, thus attracting continual research.

The key functional motif of toxins against Kv-channels is a dyad composed of a pore blocking lysine and a nearby aromatic residue. Several groups have also proposed the presence of 'basic ring' motifs to explain extreme selectivity for channel subtypes in several toxins. Since experimental validation of specificity is difficult due to numerous toxin-channel combinations, we sought to create a comprehensive database in-silico via the program HADDOCK. The Kv1.2 structure was used to construct its homologues Kv1.1 and Kv1.3, and submitted in blind-docking protocols versus ~30 toxins to isolate consensus binding modes. We find that all toxins share a small number of binding modes, classified by the identity of the residue inserted into the channel pore. HADDOCK outputs a near-native mode as the top-ranked pose in over 50% of runs, or ~90% if the top 5 is considered. Identification of other modes, e.g. associated with KCa2-channel binding, suggests that some toxins may bind to multiple targets. We also find peripheral residues have roles interacting with the channel S5P loop - confirming that the basic ring acts to discriminate Kv1 channels versus other potassium channels.

At the time of writing, we are in the process of completing the database with a hypothesis that Kv1-subtype selectivity arises from: (a) exact surface/hydrophobic matching, and (b) charge content and positioning. We ultimately aim to find an optimal scaffold for Kv1.3 and extension towards other K+-channel subtypes as more structures are published.

201-Pos

Theoretical Models of the Biological Catch-Bond

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The biological catch-bond is a fascinating and counterintuitive phenomenon, which was predicted theoretically 30 years ago. Recently, this predicted behavior has been observed in a number of protein receptor-ligand complexes. When an external force is applied to a catch-bond in an attempt to break it, either *in vivo* or *in vitro*, the bond resists breaking and becomes stronger instead. This is in contrast to ordinary slip-bonds which represent the vast majority of biological and chemical bonds and which dissociate faster when subjected to a force. This report focuses on the fundamental properties of catch-bonds and analyzes the simplest physical-chemical models to explain the experimental data. The simplicity of the theoretical treatment leads to analytic expressions for bond lifetime, concise universal representations of the experimental data, and explicit conditions required for catch-binding.

Three different model of the biological catch-bond will be discussed, including the two pathway, deformation and allosteric models. Catch-binding is a consequence of a complex potential energy landscape in a biological receptor-ligand bond. Bond lifetime can increase with force, if this force prevents dissociation through a native pathway and instead drives the system over a higher energy barrier. The lifetime can also increase if the conformations of proteins in the complex are altered by the force in a way that strengthens receptor-ligand interaction. Such bond deformation can be associated with an allosteric effect, in which a conformational change at one end of the protein propagates to the binding site located at the other end. Both experiment and simulation indicate that catch-binding is accompanied by large-scale domain opening in the receptor protein. The models are used to describe catch-binding in P-selectin/PSGL-1, FimH/mannose, actin/myosin and integrin/fibronectin complexes.

O. V. Prezhdo, Y. V. Pereverzev, ''Theoretical aspects of the biological catch-bond'', *Acc. Chem. Res.*, **42**, 693 (2009).

202-Pos

Correlation Between Functionality and Biochemical Properties in Biotin Protein Ligases

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Biotin protein ligases are a family of enzymes that catalyze biotin linkage to biotin-dependent carboxylases. In microorganisms these enzymes are functionally divided into two classes including the monofunctional class that only catalyzes biotin addition and the bifunctional class that also binds to DNA to regulate transcription. Biochemical and biophysical studies of the bifunctional Escherichia coli ligase suggest that several properties of the enzyme have evolved to support its additional regulatory role. These properties include the order of binding of multiple substrates and linkage between oligomeric state and ligand binding.

In order to test the hypothesized relationship between bifunctionality and enzymatic properties in ligases, we have carried out studies of monofunctional ligase from Pyrococcus horikoshii. Sedimentation equilibrium measurements to determine the effect of ligand binding on oligomerization indicate that the enzyme exists as a dimer regardless of liganded state. Isothermal titration calorimetry and fluorescence spectroscopy measurements of substrate binding indicate that, unlike in the E. coli enzyme, substrate binding is not ordered. Finally, thermodynamic signatures of ligand binding to the monofunctional enzyme differ significantly from those measured for the bifunctional enzyme. Combined studies of the bifunctional and monofunctional biotin ligases indicate a link between the functionality of these enzymes and their detailed biochemical characteristics.

203-Pos

Remote Regions Involved in Phosphoenolpyruvate Binding to Lactobacillus Delbreuckii Phosphofructokinase

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Phosphofructokinase from Lactobacillus delbrueckii subspecies bulgaricus (LbPFK), unlike that from Thermus thermophilus (TtPFK) and Bacillus stearothermophilus (BsPFK), exhibits weak binding affinity for the allosteric inhibitor phospho(enol)pyruvate (PEP). LbPFK has 57% sequence identity, with 75% similarity, to BsPFK and 48% identity, with 65% similarity, to TtPFK. A comparison of crystal structures between apo-LbPFK and apo-BsPFK indicates an overall conservation in structure except for the allosteric binding site. The two regions within the allosteric binding site which differ between LbPFK and other PFKs include the end of an a-helix containing residues 55-59 and a loop containing residues 211-215. Individual mutations were made in both of these regions to the corresponding residues from either TtPFK or BsPFK, with no enhancement in PEP binding. Therefore, chimeric substitutions were introduced into LbPFK in which all the residues in these regions were replaced with those from TtPFK, since TtPFK binds PEP 18,000-fold tighter when compared to LbPFK. Tt(52-61)/LbPFK and Tt(206-218)/LbPFK resulted in no enhancement in PEP binding when compared to LbPFK. They were also combined to form Tt(52-61,206-218)/LbPFK, which again exhibits similar PEP binding to LbPFK. These results indicate that the weak PEP binding in LbPFK cannot be explained as due solely to the residues which directly interact with the ligand upon binding. Another region of interest is the a-helix